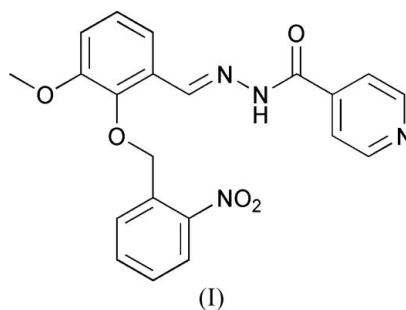


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## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
 $R$  factor = 0.057  
 $wR$  factor = 0.129  
Data-to-parameter ratio = 15.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**(*E*)-*N'*-[3-Methoxy-2-(2-nitrobenzyloxy)benzylidene]-isonicotinohydrazide**The title compound,  $\text{C}_{21}\text{H}_{18}\text{N}_4\text{O}_5$ , crystallizes with two molecules in the asymmetric unit. The conformations of these non-planar molecules are similar. Intra- and intermolecular  $\text{N}-\text{H}\cdots(\text{O},\text{N})$  hydrogen-bond systems and intermolecular  $\text{C}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bonds help to consolidate the crystal packing.Received 31 October 2006  
Accepted 8 November 2006

## Comment

There has been steady growth of interest in the structure and reactivity of Schiff bases due to their potential biological activities, such as antibacterial and antitumor (Larson & Pecoraro, 1991; Santos *et al.*, 2001). Among the large number of such compounds, isonicotinohydrazide forms a variety of Schiff bases with aldehydes, and the synthesis and crystal structures of some of these compounds, such as (*E*)-*N'*-[2-(4-chlorobenzoyloxy)benzylidene]isonicotinohydrazide (Zhang *et al.*, 2006) and (*E*)-*N'*-[4-(2-chlorobenzoyloxy)benzylidene]isonicotinohydrazide (Zhao *et al.*, 2006), have been reported. The title compound, (I) (Fig. 1), is another example of this family.The asymmetric unit of (I) consists of two independent molecules, which are similar to each other. All the bond lengths and angles are within normal ranges (Allen *et al.*, 1987). In molecule 1, the *o*-vanillin group (C7–C13/O2/O3) is planar, with an r.m.s. deviation,  $\delta$ , from the mean plane of 0.0282 Å. It makes dihedral angles of 44.92 (10) and 85.36 (7)° with the pyridine ring (C1–C5/N1) and the nitrobenzene ring (C16–C21), respectively. In molecule 2, the *o*-vanillin group (C28–C34/O7/O8) is also planar, with  $\delta = 0.0332$  Å, and it makes dihedral angles of 61.98 (9) and 89.47 (7)° with the pyridine ring (C22–C26/N5) and the nitrobenzene ring (C37–C42), respectively.Bifurcated  $\text{N}-\text{H}\cdots(\text{O},\text{N})$  intra- and intermolecular hydrogen-bond systems are found in (I) (Table 1 and Fig. 2), which help to consolidate the crystal packing. The crystal structure of (I) also contains four weak intermolecular hydrogen bonds, two  $\text{C}-\text{H}\cdots\text{N}$  and two  $\text{C}-\text{H}\cdots\text{O}$ .

## Experimental

An anhydrous ethanol solution (50 ml) of 2-(2-nitrobenzyloxy)-3-methoxybenzaldehyde (2.87 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of isonicotinohydrazide (1.37 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen, giving a yellow precipitate. The product was isolated and recrystallized from acetonitrile, and then dried in a vacuum to give the pure compound in 82% yield. Yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an acetonitrile solution.

### Crystal data

$C_{21}H_{18}N_4O_5$	$Z = 8$
$M_r = 406.39$	$D_x = 1.308 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 7.8965 (17) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$b = 26.120 (6) \text{ \AA}$	$T = 294 (2) \text{ K}$
$c = 20.139 (4) \text{ \AA}$	Block, yellow
$\beta = 96.355 (4)^\circ$	$0.24 \times 0.20 \times 0.12 \text{ mm}$
$V = 4128.3 (15) \text{ \AA}^3$	

### Data collection

Bruker SMART APEX CCD area-detector diffractometer	23178 measured reflections
$\varphi$ and $\omega$ scans	8411 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	3364 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.958$ , $T_{\max} = 0.989$	$R_{\text{int}} = 0.082$
	$\theta_{\text{max}} = 26.4^\circ$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0359P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.057$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.129$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.01$	$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$
8411 reflections	$\Delta\rho_{\text{min}} = -0.22 \text{ e \AA}^{-3}$
544 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: 0.0021 (2)

**Table 1**

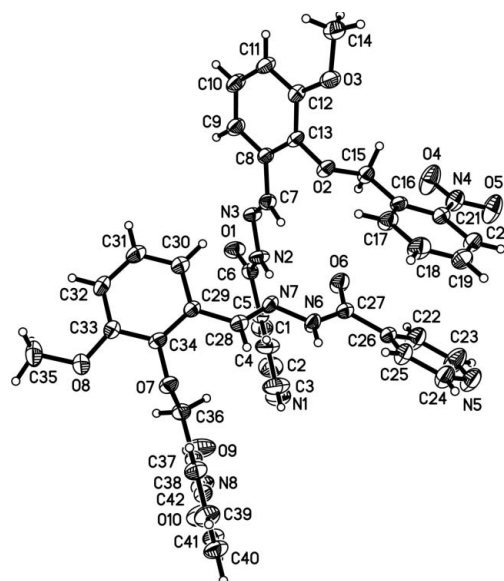
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H2A\cdots O6$	0.86	2.32	2.957 (3)	131
$N2-H2A\cdots N7$	0.86	2.54	3.314 (3)	150
$N6-H6\cdots O1^i$	0.86	2.34	2.894 (3)	123
$N6-H6\cdots N3^i$	0.86	2.53	3.365 (3)	165
$C25-H25\cdots N3^i$	0.93	2.55	3.382 (4)	150
$C40-H40\cdots O9^i$	0.93	2.56	3.448 (4)	159
$C32-H32\cdots N5^{ii}$	0.93	2.45	3.299 (4)	151
$C14-H14C\cdots O10^{ii}$	0.96	2.54	3.445 (4)	157

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x + 1, -y + \frac{1}{2}, z + \frac{1}{2}$ .

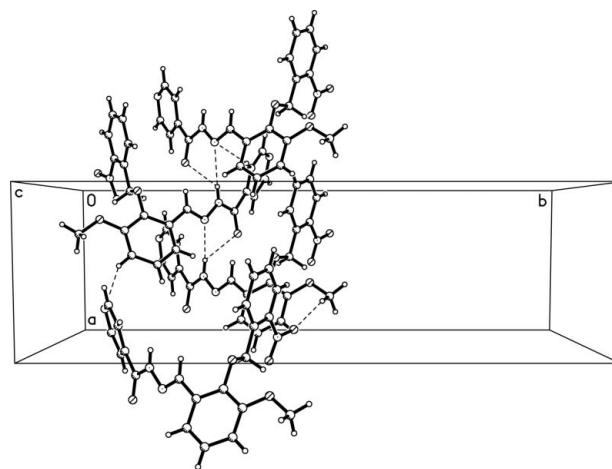
The H atoms were included in calculated positions and refined using a riding model approximation. Constrained C—H and N—H bond lengths and isotropic  $U$  parameters:  $0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for  $\text{Csp}^2\text{—H}$ ;  $0.97 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for methylene C—H;  $0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl C—H;  $0.86 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$  for imino N—H.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.



**Figure 1**

The asymmetric unit of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.



**Figure 2**

Partial packing diagram for (I), with hydrogen bonds drawn as dashed lines.

The project was supported by the Foundation of the Education Department of Hebei Province (grant No. 606022).

## References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bruker (1999). SMART (Version 5.0) and SAINT (Version 4.0) for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Larson, E. J. & Pecoraro, V. L. (1991). *J. Am. Chem. Soc.* **113**, 3810–3818.
- Santos, M. L. P., Bagatin, I. A., Pereira, E. M. & Ferreira, A. M. D. C. (2001). *J. Chem. Soc. Dalton Trans.* pp. 838–844.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Zhang, Q.-Z., Zhao, Y.-L., Chen, X. & Yu, M. (2006). *Acta Cryst. E62*, o4835–o4836.
- Zhao, Y.-L., Zhang, Q.-Z., Chen, X. & Yu, M. (2006). *Acta Cryst. E62*, o4928–o4929.